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VIT[®]
Vellore Institute of Technology
(Deemed to be University under section 3 of UGC Act, 1956)

SCHOOL OF ADVANCED SCIENCES
Department of Chemistry
Winter Semester 2023-24
Continuous Assessment Test – I

Course Code: BCHY101L

Duration : 90 Minutes

Slot: E1+TE1

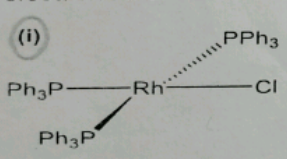
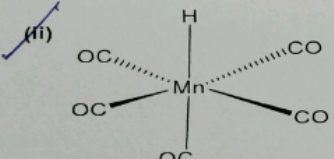
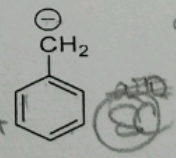


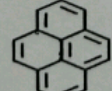

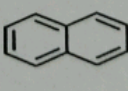
Course Name: Engineering Chemistry

Max. Marks : 50

Class Numbers: VI2023240502637; 2638; 2640; 2641; 2646; 2649 Faculty Names:

Dr. Buvaneswari G; Dr. Nawaz Khan F; Dr. Rajagopal D; Dr. Sasikumar S; Dr. Napoleon A.A;

Dr. Madhvesh Pathak

Q N	Answer <u>ALL</u> the questions (5 x 10 = 50 Marks)	Marks	CO	BL
1	Comment on the hybridisation, magnetic character, geometry and calculate the magnetic moment for the complexes with a suitable diagram. i) $[\text{Fe}(\text{CN})_6]^{4-}$ ii) $[\text{CuCl}_4]^{2-}$	10	CO1	BL1
2	(a) Determine the CFSE value for the following complexes i) $[\text{Cr}(\text{CN})_6]^{3-}$ ii) $[\text{CoF}_6]^{3-}$ (b) Describe the structural features of Hemoglobin and give the role of proximal and distal histidine with illustration.	(5 + 5)	CO1	BL3
3	(a) Predict the stability of the following organometallic compound using 18-electron rule. (i)  (ii)  (b) Give the method of preparation of an antipyretic drug paracetamol and explain the role of acetic anhydride with necessary steps involved.	(5 + 5)	CO1	BL3
4	(a) Explain the stability of <u>primary, secondary and tertiary carbocations</u> using <u>hyperconjugation</u> and <u>inductive effect</u> with suitable examples. (b) Write down the stability order for following four different carbanions and justify your answer. (i) $\text{R}_3\text{C}-\text{CH}_2^-$ (ii) $\text{R}_2\text{C}=\text{CH}^-$ (iii) $\text{RC}\equiv\text{C}^-$ (iv) 	(5 + 5)	CO1	BL3
5	Identify the following compounds as aromatic, antiaromatic, nonaromatic and justify your answer.  (i)  (ii)  (iii)  (iv)  (v)	10	CO1	BL2