



Final Assessment Test – November 2025

Course: TBIT302L - Bioprocess Engineering

Class NBR(s): 1026 / 1029

Time: Three Hours

Slot: C1+TC1

Max. Marks: 100

- KEEPING MOBILE PHONE/ANY ELECTRONIC GADGETS, EVEN IN 'OFF' POSITION IS TREATED AS EXAM MALPRACTICE
- DON'T WRITE ANYTHING ON THE QUESTION PAPER

COs	CO Statements
CO1	Analyze media design and optimization process.
CO2	Evaluate oxygen transfer rate in bioreactors and mode of sterilization process.
CO3	Create mathematical models for microbial growth.
CO4	Apply mass transfer principles to solve biological problems.
CO5	Relate importance of metabolic stoichiometry in microbial growth and product formation

BL – Blooms Taxonomy Level (1 – Remember, 2 – Understand, 3 – Apply, 4 – Analyse, 5 – Evaluate, 6 – Create)

Answer ALL Questions

(10 X 10 = 100 Marks)

- | | | | |
|----|---|-----|-----|
| 1. | Provide a clear and well-labelled diagram of a mixed flow bioreactor, along with a detailed explanation of its benefits and drawbacks. | CO1 | BL1 |
| 2. | Describe with a well-labelled diagram of photobioreactors, accompanied by a detailed description of their strength and weaknesses. | CO1 | BL1 |
| 3. | Explain the working principle of a fed-batch bioreactor, including its concentration profile, and discuss its merits and demerits. | CO1 | BL1 |
| 4. | Outline the classical and statistical approaches used in media optimization, highlighting their principles and comparative advantages | CO3 | BL2 |
| 5. | Explain the process of continuous heat sterilization of liquids, highlighting its operational principles and equipment involved. | CO2 | BL1 |
| 6. | Illustrate the various phases of cell growth in batch cultures with respect to the specific growth rate. | CO3 | BL2 |
| 7. | Explain the Luedeking-Piret model for microbial product formation, including its mathematical formulation and significance in bioprocess engineering. | CO5 | BL2 |
| 8. | Using the Monod kinetics framework, explain the methods employed to determine $\mu_{\max}$ and K_s . | CO3 | BL2 |

- 9.a) Medium at a flow rate of $2 \text{ m}^3/\text{h}$ is to be sterilized by heat exchange with stream in a continuous sterilizer. The liquid contains bacterial spores at concentration of $5 \times 10^{12} \text{ m}^{-3}$; the activation energy and Arrhenius constant for thermal destruction of these contaminants are 283 kJ/gmol and $5.7 \times 10^{39} \text{ h}^{-1}$ respectively. A contamination risk of one organism surviving every 60 days operation is considered acceptable. The sterilizer pipe has a inner diameter of 0.1 m ; the length of the holding section is 24 m . The density of the medium is 1000 kg/m^3 and the viscosity is 3.6 kg/ (m h) . Find ratio of N_2/N_1 , Peclet number, specific death constant (K_d), What sterilizing temperature is required.

Data: Value for $D_z/uD=0.65$, Damkohler number (Da) is 42 from graph.

OR

- 9.b) A 45 L stirred fermenter containing *B.thuringiensis* culture at 30°C is used for the production of microbial insecticide, find K_{ia} using a dynamic method. Airflow shuts off for a few minutes and dissolved oxygen levels drop; the air supply is reconnected. When a steady state is reached, the dissolved oxygen concentration is 78% air saturation. The following results are obtained.

Time (s)	5	15
Oxygen conc (% air saturation)	50	66

- 10.a) Production of a single cell protein from hexadecane by the following reaction equation:



Where $\text{CH}_{1.66}\text{N}_{0.27}\text{O}_{0.20}$ represent the biomass. IF $RQ = 0.43$, determine the stoichiometric coefficients.

OR

- 10.b) *Klebsiella aerogenes* ($\text{CH}_{1.75}\text{O}_{0.43}\text{N}_{0.22}$) is produced from glycerol in aerobic culture with ammonia as nitrogen source. The biomass contains 8% ash, 0.40 g biomass is produced for each g glycerol consumed, and no major metabolic products are formed. What is the oxygen requirement for this culture in mass terms? The stoichiometry equation for anaerobic growth of *Klebsiella aerogenes* is given by
- $$\text{C}_3\text{H}_8\text{O}_3 + a\text{O}_2 + b\text{NH}_3 \rightarrow c\text{CH}_{1.75}\text{O}_{0.43}\text{N}_{0.22} + d\text{CO}_2 + e\text{H}_2\text{O}$$

U/K/TY

CO4 BL3

CO4 BL3

CO5 BL3

CO5 BL3